

LECTURE 07: TECHNICAL BASIS FOR OPTICAL/LASER-BASED FLOW DIAGNOSTICS

Dr. Hui Hu

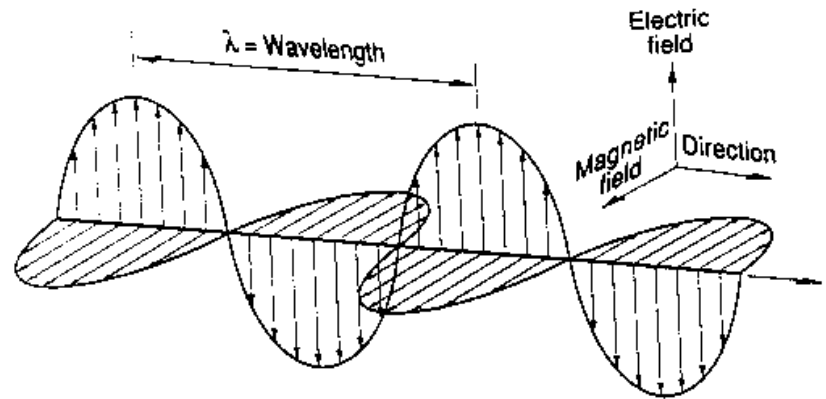
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THE NATURE OF LIGHT

- Classical electromagnetic theory: light is radiation that propagates through vacuum in free space in the form of electromagnetic waves, both oscillating transversely to the direction of wave propagation and normal to each other.

$$E_y(x, t) = E_{y0} \sin 2\pi\left(\frac{x}{\lambda} - \frac{t}{T}\right)$$

$$B_z(x, t) = B_{z0} \sin 2\pi\left(\frac{x}{\lambda} - \frac{t}{T}\right)$$



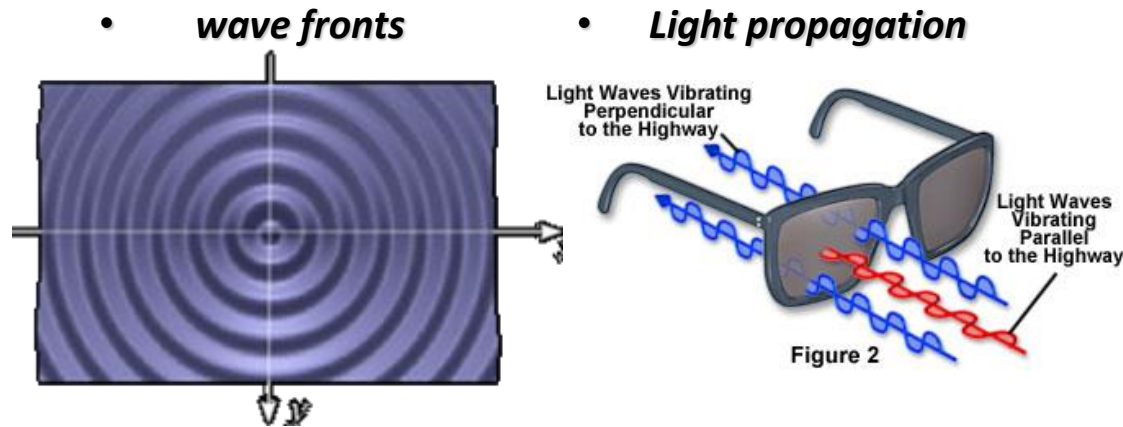
- λ : is wavelength
- T : is the period of the oscillation
- f : The reciprocal of the period, is called frequency, $f = 1/T$

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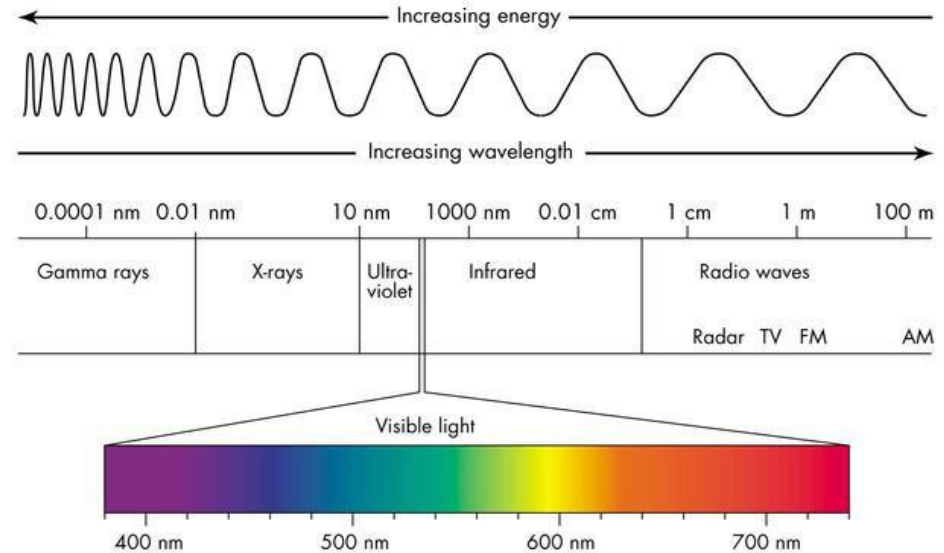
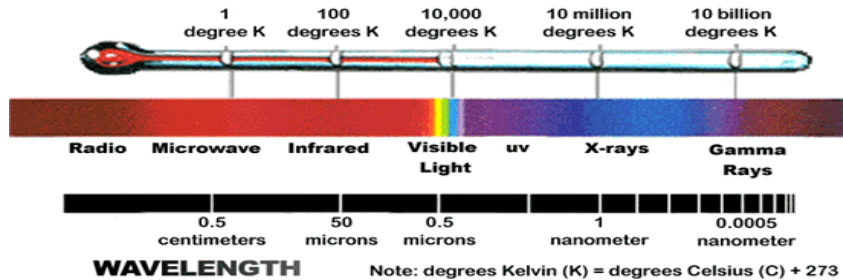
NATURE OF LIGHT

□ THE NATURE OF LIGHT- AS ELECTROMAGNETIC WAVES

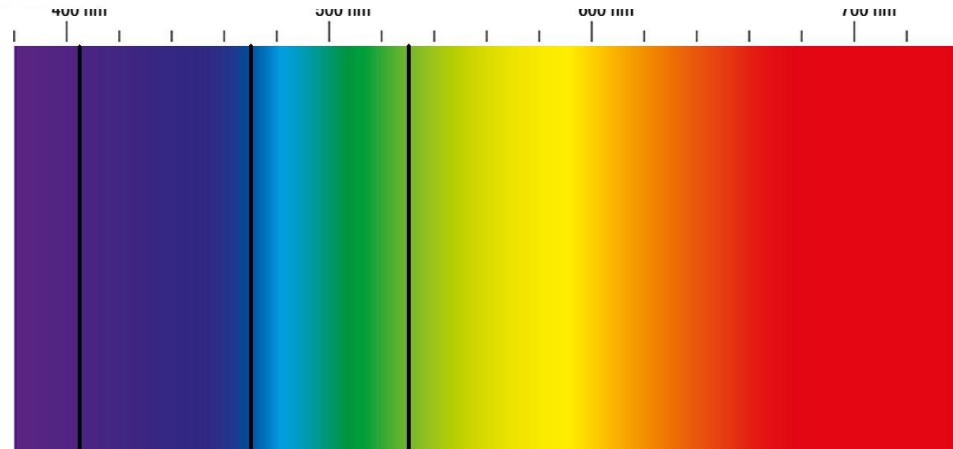
- Light propagation velocity, $V = f \lambda$
- Light propagation velocity in Vacuum, $C = 2.998 \times 10^8 \text{ m/s}$
- Wave front: the locus of all points along the different paths that have the same phase.
- If all the wave fronts are plane, then, the light is considered to be a plane wave.
- If all the wave fronts are spherical or cylindrical, then, the light is considered to be a spherical or cylindrical wave.
- Light propagation is associated with electric and magnetic fields.
- The polarization is associated with the orientation of the plane of the plane of oscillation of the electric field.
- Concepts of linearly polarized light, elliptically polarized light and circularly polarized light, unpolarized or randomly polarized light.



THE NATURE OF LIGHT- AS ELECTROMAGNETIC WAVES

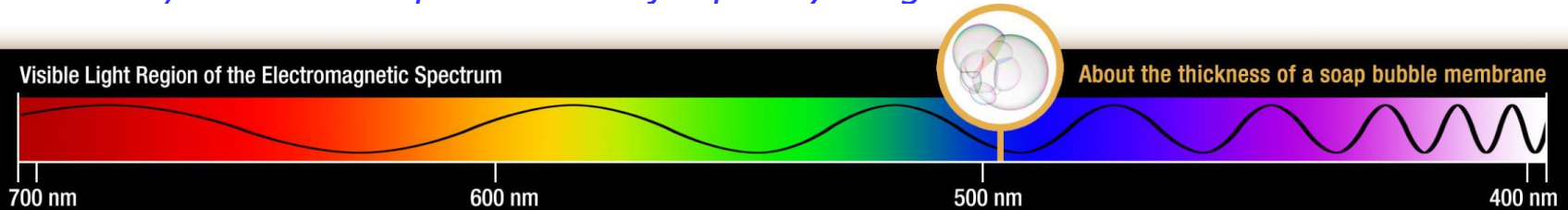


RADIATION TYPE	WAVELENGTH RANGE
Cosmic rays	$\lambda < 10^{-4} \text{ nm}$
Gamma rays	$10^{-4} \text{ nm} < \lambda < 10^{-1} \text{ nm}$
X-rays	$10^{-2} \text{ nm} < \lambda < 10^2 \text{ nm}$
Disinfecting radiation	$10 \text{ nm} < \lambda < 380 \text{ nm}$
Visible light	$380 \text{ nm} < \lambda < 750 \text{ nm}$
Space heating	$750 \text{ nm} < \lambda < 10^7 \text{ nm}$
Microwaves	$10^6 \text{ nm} < \lambda < 10^9 \text{ nm}$
Radar	$10^7 \text{ nm} < \lambda < 10^9 \text{ nm}$
Radio and Television	$10^8 \text{ nm} < \lambda < 10^{13} \text{ nm}$
Electrical power waves	$10^{14} \text{ nm} < \lambda < 10^{17} \text{ nm}$

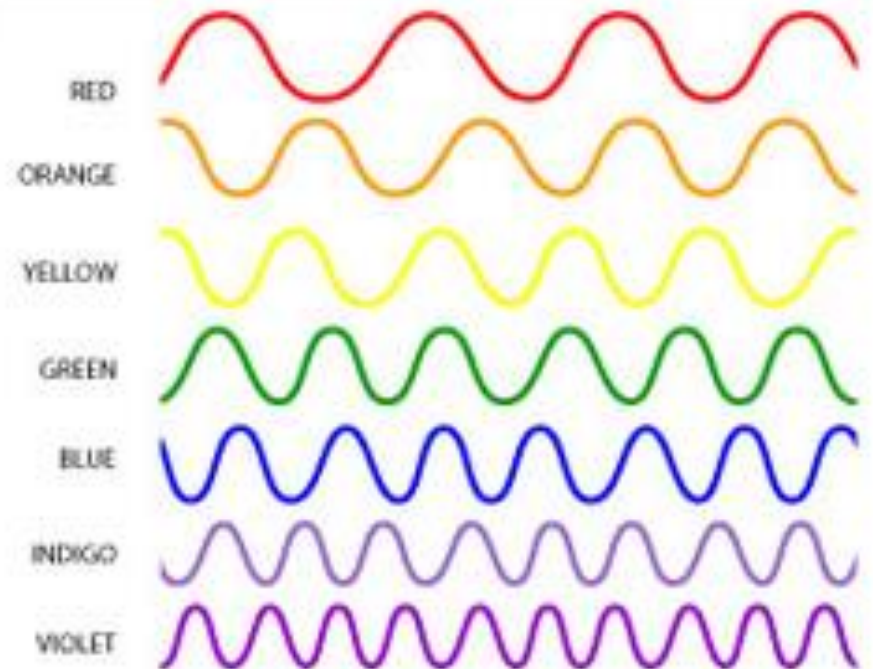


☐ THE NATURE OF LIGHT- AS ELECTROMAGNETIC WAVES

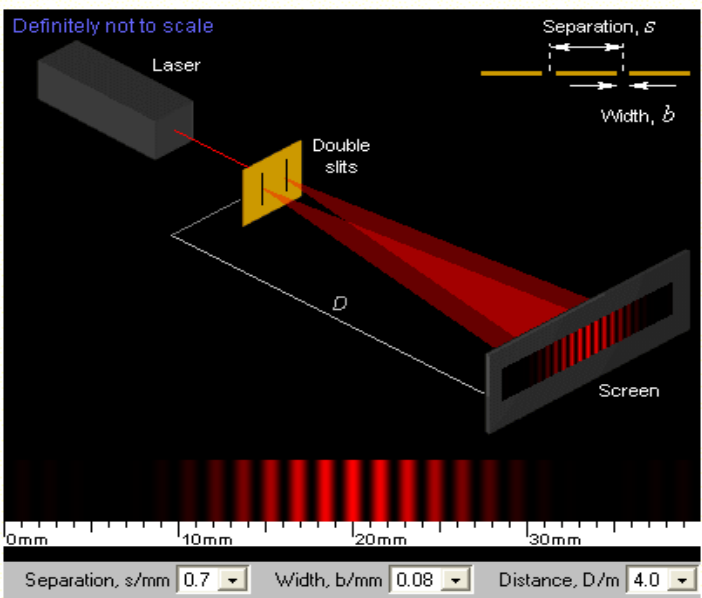
- The colors: visible light consists of radiation with wavelength in the range of 380~750nm ($1\text{nm}=10^{-9}\text{m}$) which corresponds to the frequency range between 4.0×10^{15} to 7.9×10^{15} Hz.*



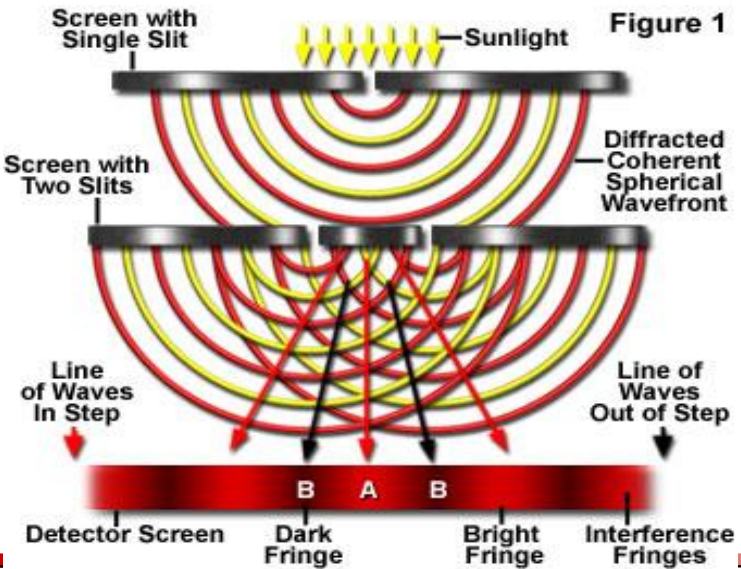
COLOR	WAVELENGTH RANGE
Ultraviolet (UV)	$0.85 \text{ nm} < \lambda < 380 \text{ nm}$
Violet	$380 \text{ nm} < \lambda < 424 \text{ nm}$
Blue	$424 \text{ nm} < \lambda < 491 \text{ nm}$
Green	$491 \text{ nm} < \lambda < 575 \text{ nm}$
Yellow	$575 \text{ nm} < \lambda < 585 \text{ nm}$
Orange	$585 \text{ nm} < \lambda < 647 \text{ nm}$
Red	$647 \text{ nm} < \lambda < 750 \text{ nm}$
infrared	$750 \text{ nm} < \lambda < 1000 \text{ nm}$



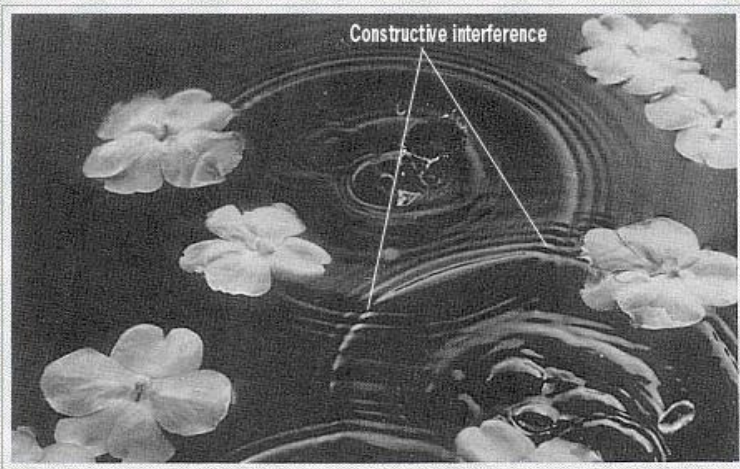
☐ LIGHT AS WAVES - INTERFERENCE OF LIGHT WAVES



Thomas Young's Double Slit Experiment

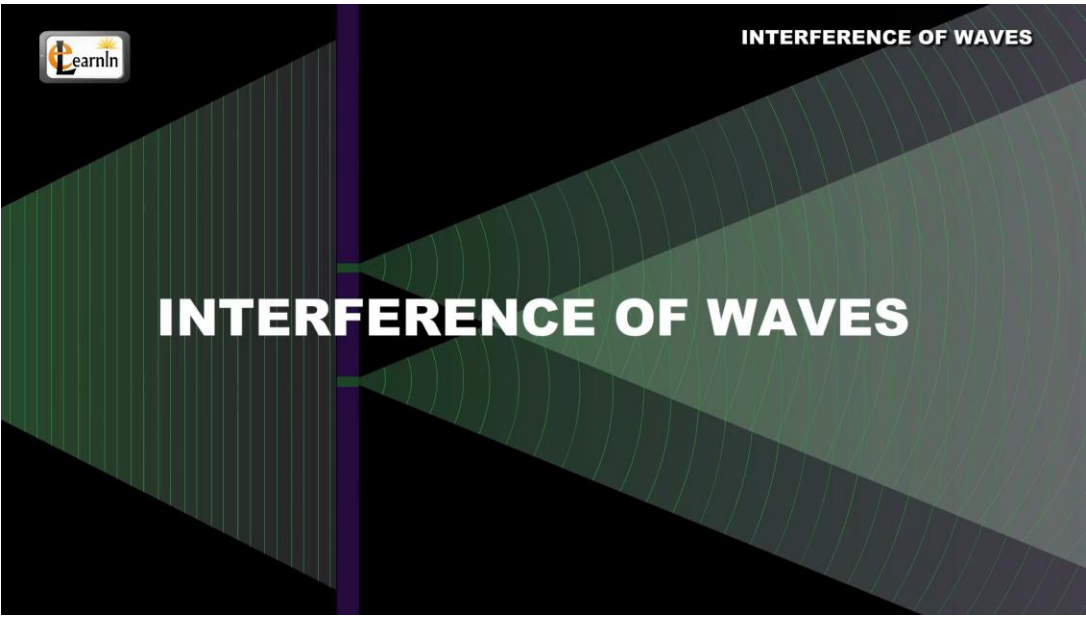


Inteference of Waves from Two Sources



In some places the water wavefronts are in phase (bright spots).

In other places the fronts overlap with peak and valley and interfere destructively (darker spots).

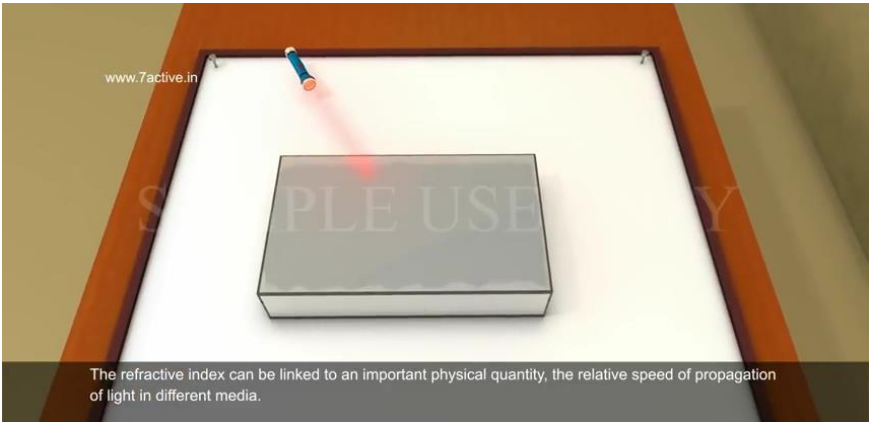


LIGHT PROPAGATION THROUGH MEDIA

- Refractive index:** $n = c / v = \frac{\lambda_0}{\lambda} > 1$

Index of refraction of a material generally increasing slightly with decreasing wavelength of the light. Such phenomena is called dispersion.

$$n = \sqrt{\frac{1 + 2K_L\rho}{1 - K_L\rho}}$$



<i>Gas</i>	<i>n</i>	<i>Liquid</i>	<i>n</i>	<i>Solid</i>	<i>n</i>
<i>Air</i>	<i>1.00029</i>	<i>Water</i>	<i>1.333</i>	<i>Fused quartz</i>	<i>1.46</i>
<i>He</i>	<i>1.00036</i>	<i>Ethyl alcohol</i>	<i>1.361</i>	<i>Pyrex glass</i>	<i>1.47</i>
<i>CO₂</i>	<i>1.00045</i>	<i>Turpentine</i>	<i>1.472</i>	<i>Crown glass</i>	<i>1.52</i>
<i>H₂</i>	<i>1.00013</i>	<i>Benzene</i>	<i>1.501</i>	<i>Flint glass</i>	<i>1.57~1.89</i>
				<i>Plexiglas</i>	<i>1.51</i>
				<i>Lexan</i>	<i>1.58</i>
				<i>Polystyrene</i>	<i>1.59</i>
	<i>589nm</i>			<i>sapphire</i>	<i>1.77</i>
				<i>zircon</i>	<i>1.92</i>
				<i>Diamond</i>	<i>2.42</i>

□ LIGHT PROPAGATION THROUGH MEDIA

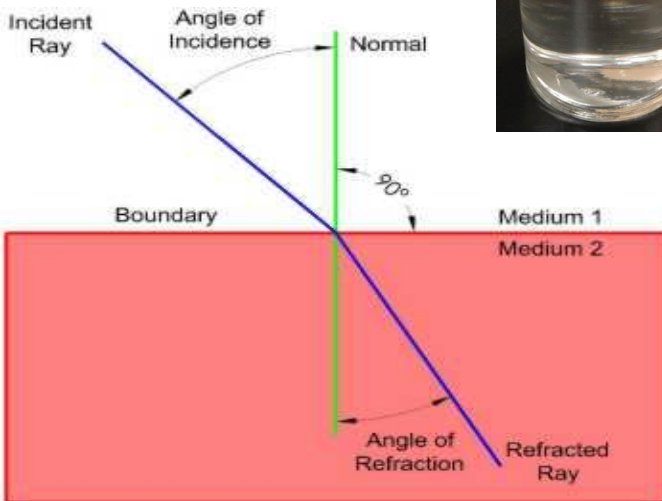
- **Refraction:**

- When light propagates through a homogenous medium, its path would be straight, whereas, if the medium is non-homogeneous or if the light across from one medium to another, the path may change direction gradually or abruptly. The change of light propagation direction is called refraction.

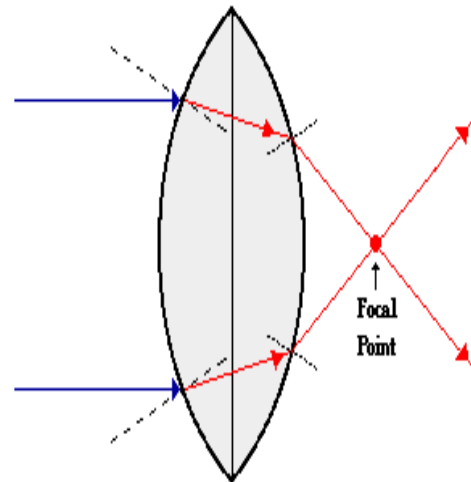
$$n = c / v = \frac{\lambda_0}{\lambda} > 1 \quad \frac{\sin \varphi_1}{\sin \varphi_2} = \frac{n_2}{n_1}$$

- Optically denser medium ($n_1 < n_2$)

- **Snell's Law:**

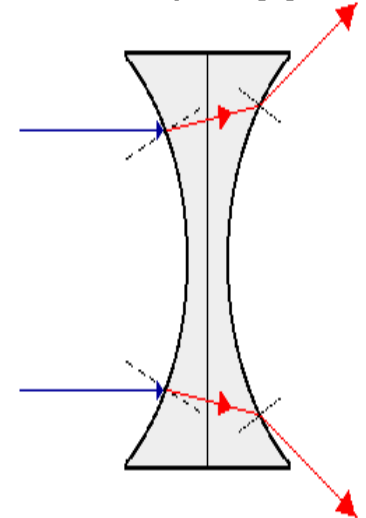


Refraction by a Converging Lens



Incident rays which travel parallel to the principal axis will refract through the lens and converge to a point.

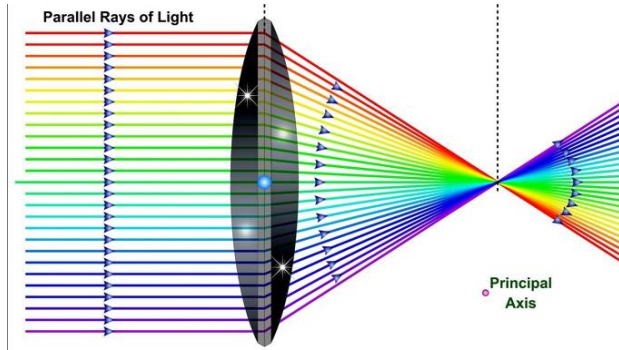
Refraction by a Diverging Lens



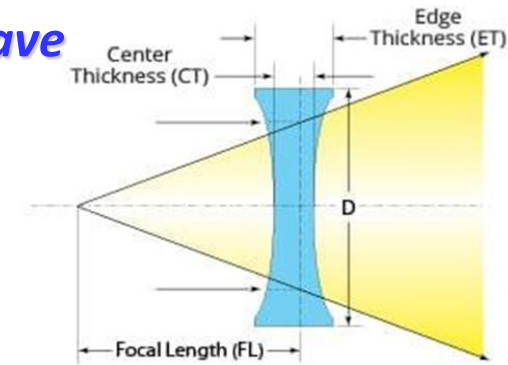
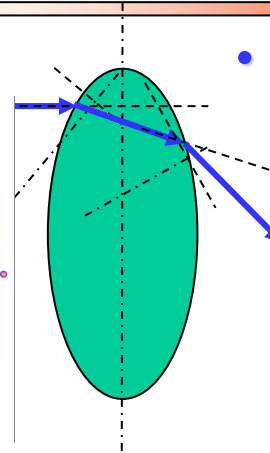
Incident rays traveling parallel to the principal axis will refract through the lens and diverge, never intersecting.

LIGHT PROPAGATE THROUGH MEDIA

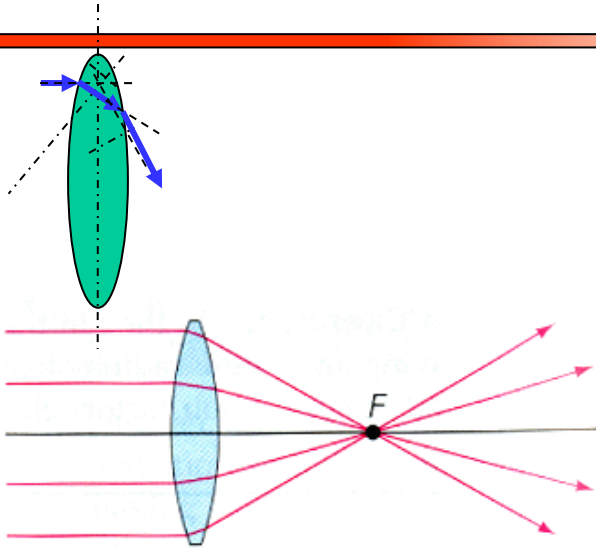
- **Convex lens**



- **Concave lens**

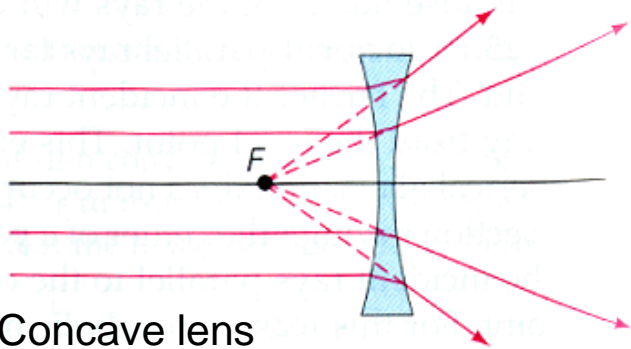


❏ OPTICAL LENSES

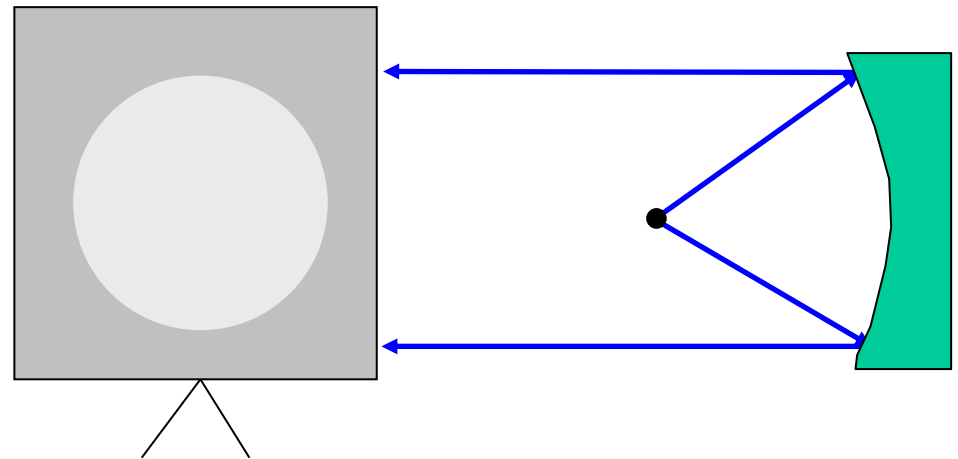
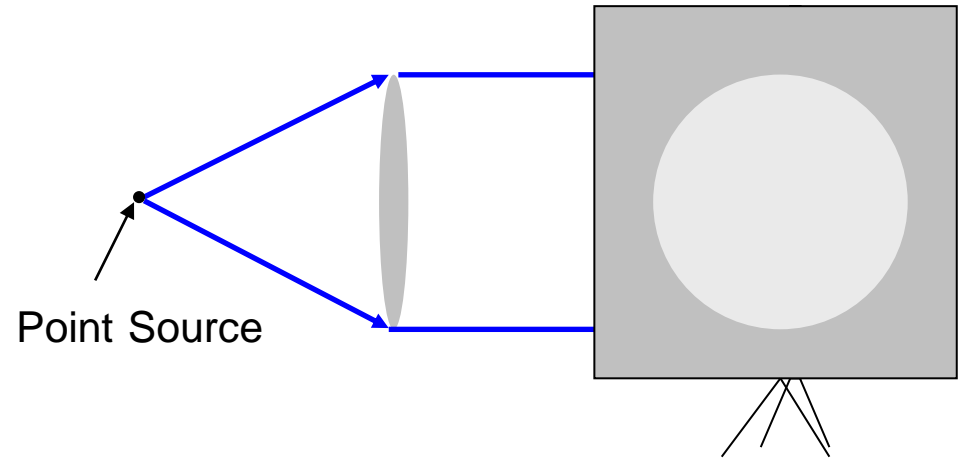


Converging lens

Convex lens



Concave lens



□ LIGHT PROPAGATION THROUGH MEDIA

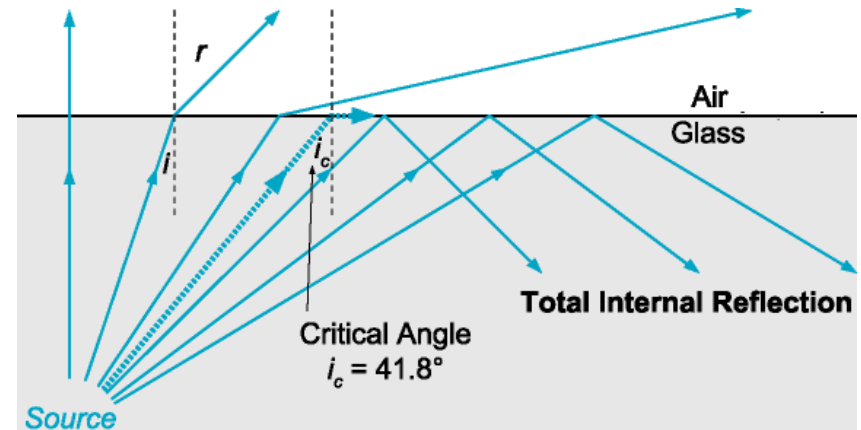
• Total refraction

$$\frac{\sin \phi_1}{\sin \phi_2} = \frac{n_2}{n_1}$$

$$\sin \phi_2 = \frac{n_1}{n_2} \sin \phi_1$$

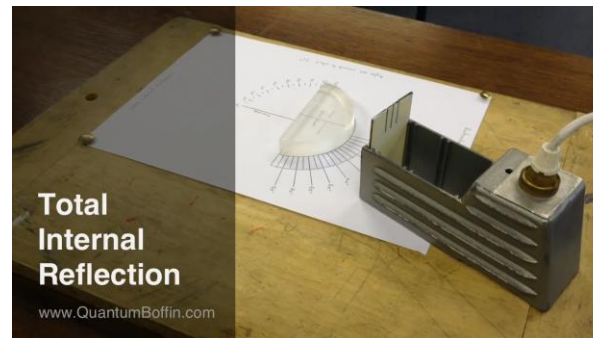
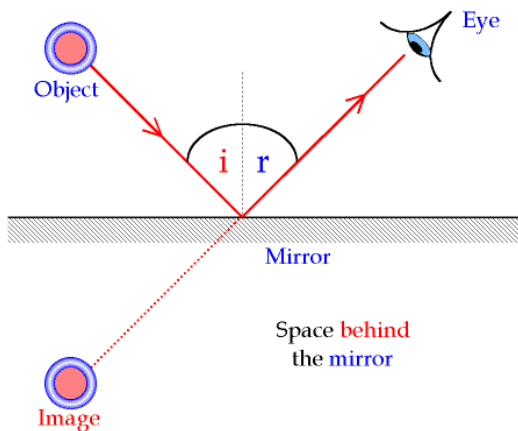
$$n_1 > n_2 \Rightarrow \phi_2 > \phi_1$$

$$\phi_{2\max} > \pi/2 \Rightarrow \phi_{1\text{cri}} = \sin^{-1}(n_2/n_1)$$



Refraction and total internal reflection.

• Reflection



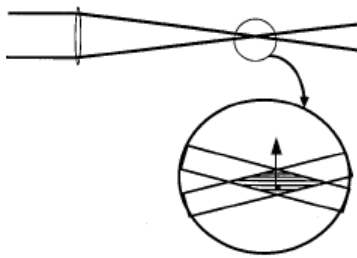
Total Internal Reflection in Water

starring Daniel

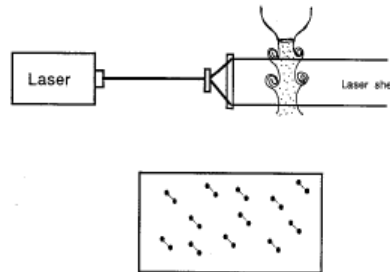
How does refractive-index-matching help?

- Optical techniques avoid disturbing the flow to be measured
- Typical approaches are LDV, PIV, PTV, flow visualization, PLIF, etc.

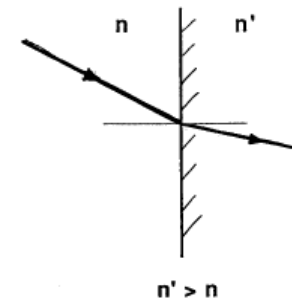
Laser Doppler Velocimetry



Particle Image Velocimetry

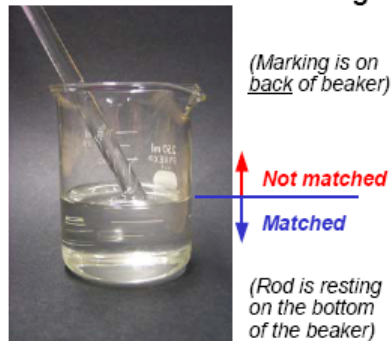


Snell's Law

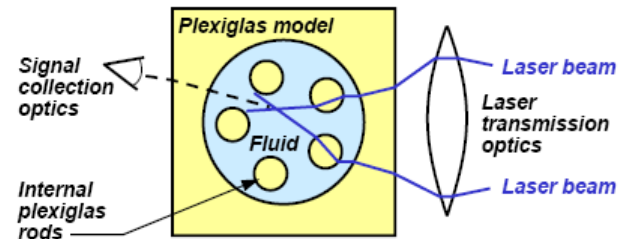


- Unless the refractive indices are matched, the view may be distorted or impossible even with "transparent" materials and position measurements may be incorrect

Example of application of refractive-index-matching



Refractive index not matched



❑ THE OPTICAL INDEX OF REFLECTION MATCHING APPROACH

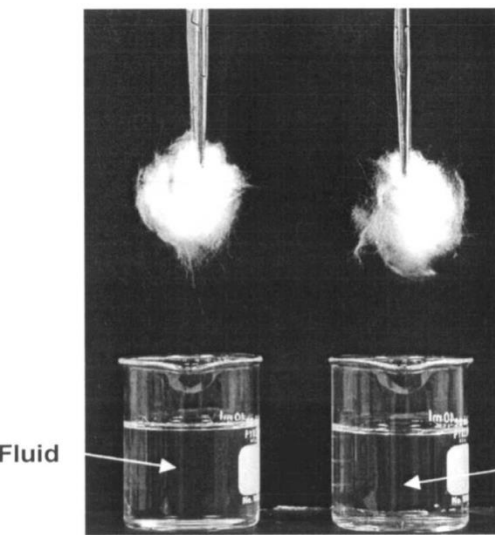


Figure 3. Wool in air

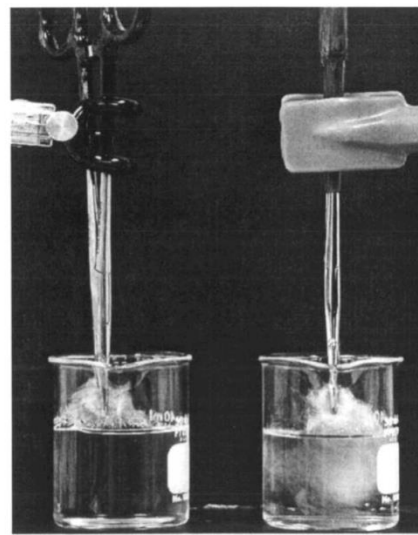


Figure 4. Half submerged

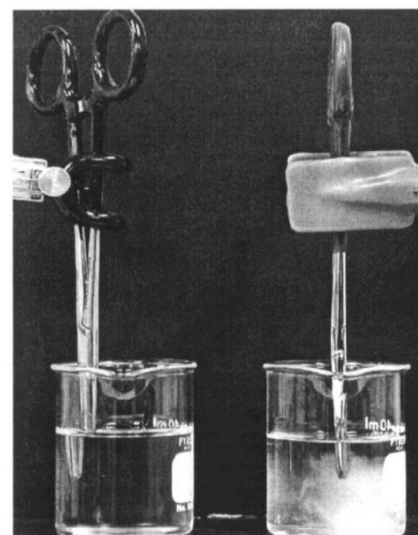
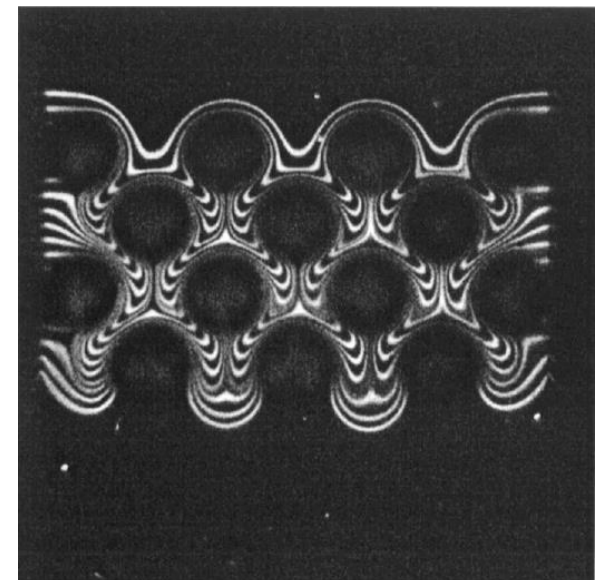
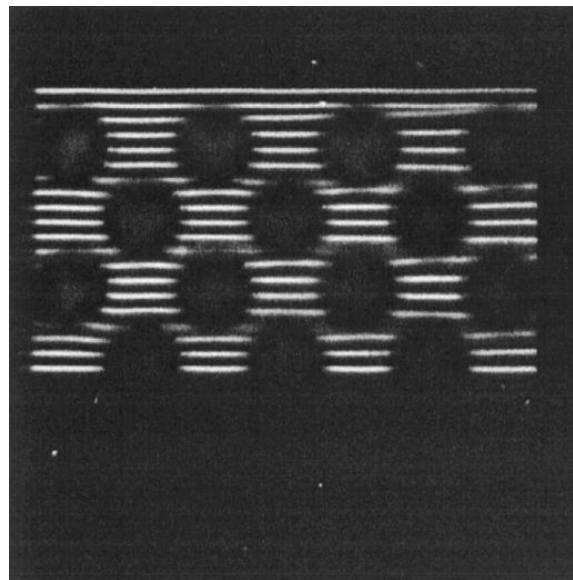
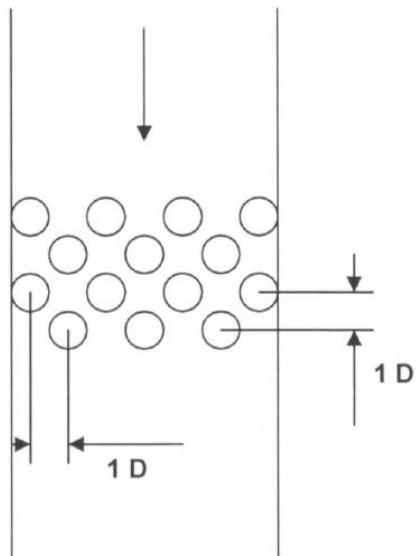


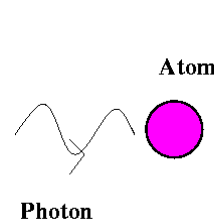
Figure 5. Totally submerged



☐ THE NATURE OF LIGHT – AS PHOTONS

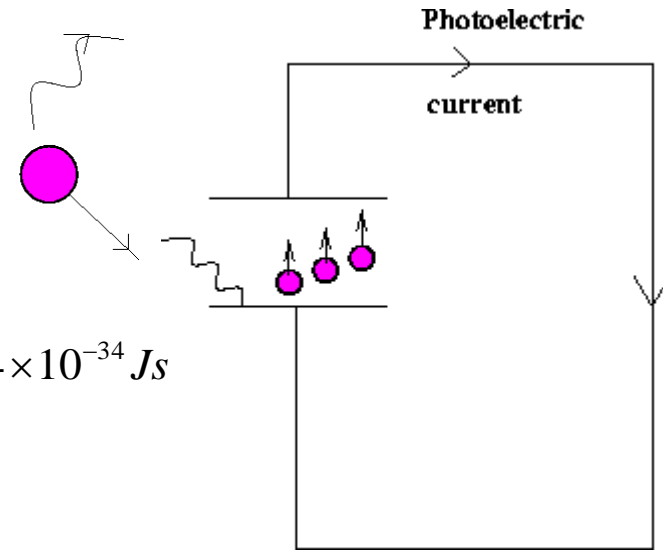
Photon scattering:

- one finds experimentally that the frequency of the scattered wave is changed, which does not come out of a wave picture of light. However, when the light is viewed as a photon with energy proportional to the associated light wave, excellent agreement with experiment is found.



$$\varepsilon = h\nu$$

$$\text{Planck const} = 6.624 \times 10^{-34} \text{ Js}$$



The photoelectric effect:

- When light shines on a metal plate, electrons are ejected. These electrons are accelerated to a nearby plate by an external potential difference, and a photoelectric current is established, as below
- The photons hit an electron in the metal, giving up energy. If photon energy is sufficient to free the electron, it is accelerated towards the other side; hence, a flow of charges (current).
- The photoelectric current depends critically on the frequency of the light. This is a feature of the energy that the electrons gain when struck by the light.
- **The explanation was first given by Einstein and won him the Nobel Prize.**

Amazing
Moments
in
Science

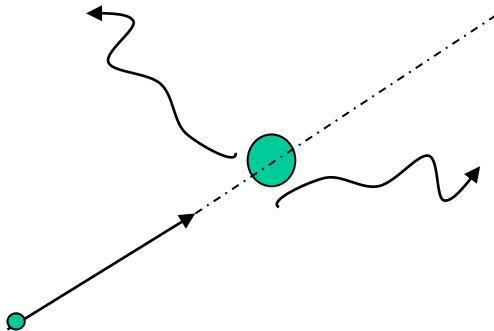
by:

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❏ Light Scattering

- **Scattering**
 - *Scattering is a general physical process whereby some forms of radiation, such as light, are forced to deviate from a straight trajectory by one or more localized non-uniformities in the medium through which it passes.*
- **Elastic Scattering**
 - *Excited electron or atoms emits a photon with exact the same frequency as the incident one.*
- **Inelastic scattering**
 - *Excited electron or atoms emits a photon with a frequency different from the incident one.*

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ELASTIC SCATTERING

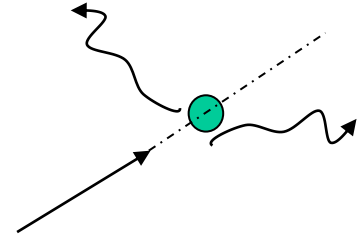
Rayleigh Scattering

- Light scattering from particles that are smaller than 1/15 of the incident light wavelength ($d < \lambda/15$).
- Efficiency of the scattering from a particle is expressed in terms of scattering cross section.

$$\sigma_R = \sigma_T \left(\frac{\lambda_0}{\lambda} \right)^2$$

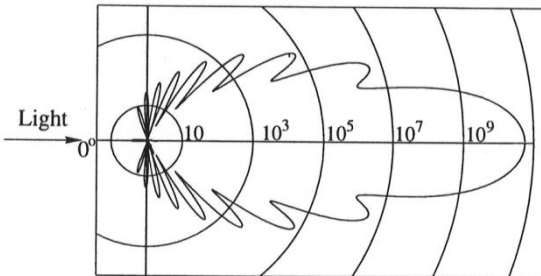
$$\sigma_T = 6.65 \times 10^{-29} m^2$$

λ_0 is the characteristic wavelength of the atom.

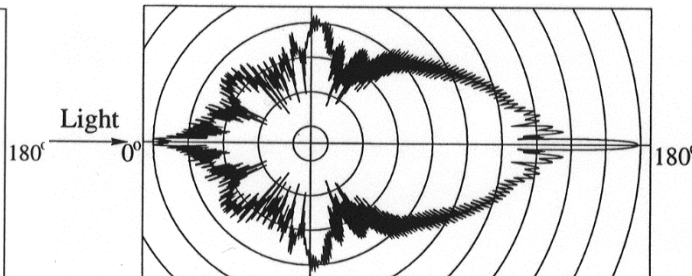


Mie Scattering

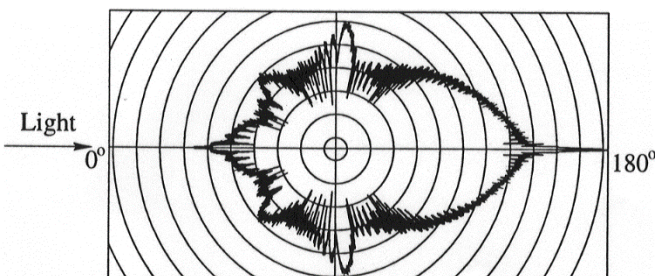
- Light scattering from particles with its size close on bigger than the incident light wavelength ($d > \lambda$).
- Conservation of polarization direction
- Angle dependent
 - Forward scattering
 - Back scattering



(a). $d=1\mu m$



(b). $d=10\mu m$



(c). $d=30\mu m$

❏ Inelastic Scattering

Raman Scattering

- Inelastic scattering from molecules.
- Chance to occur is about $10^{-5} \sim 10^{-2}$ of times lower than the Rayleigh scattering
- Scattering cross section is several orders smaller than the Rayleigh scattering
- Stoke transition : the energy of the emitted photon is higher than the absorbed photo.
- Anti-stoke transition: the energy of the emitted photon is lower than the absorbed photo.
- Time between the absorption and emission: 10^{-14} s.
- Anti-stokes line will be stronger when the temperature is low.

Scattering of light by molecules

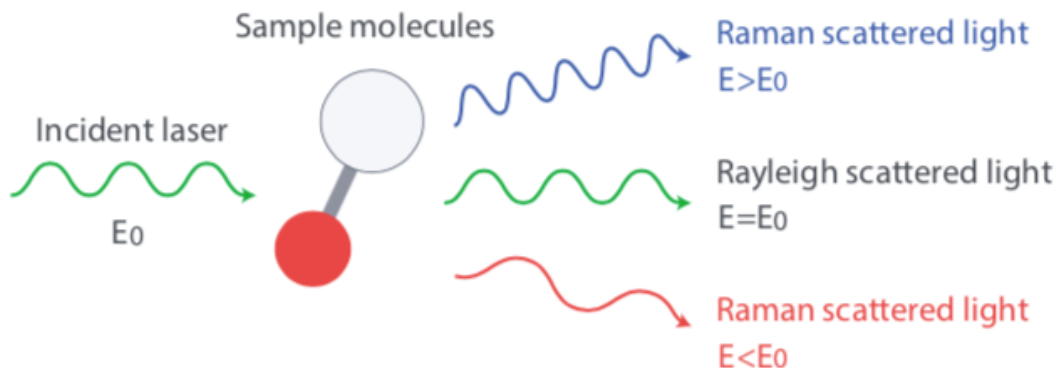


Table 5.5. Wavelengths of Raman-scattered radiation for some common molecules in air at standard atmospheric pressure and a temperature of 295 K; excitation was provided by a ruby laser

Molecule	Anti-Stokes line (nm)	Stokes line (nm)
Incident light	694.30 (Rayleigh line)	
CO ₂	638.23	761.17
CO ₂	637.92	762.23
CO ₂	633.25	768.37
CO ₂	632.42	769.61
O ₂	626.57	778.44
O ₂ ⁺	615.50	796.23
NO	614.26	798.31
CO	604.34	815.73
N ₂ ⁺	603.19	817.82
N ₂	597.57	828.39
CH ₄	577.45	870.46
H ₂	538.67	976.41

FLUORESCENCE AND PHOSPHORESCENCE

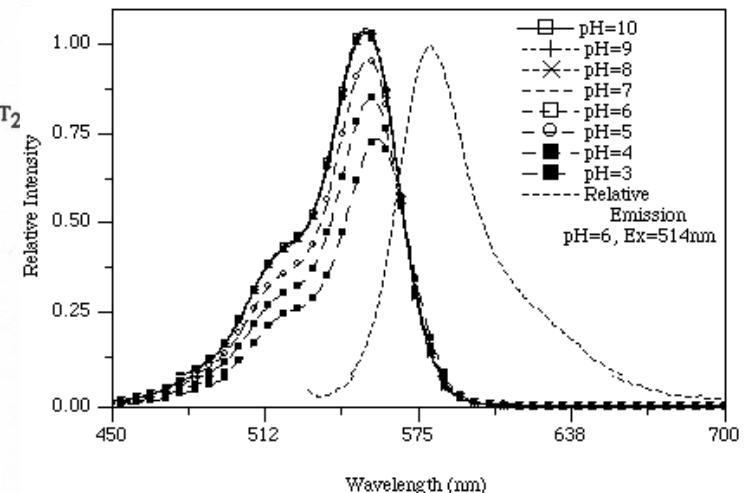
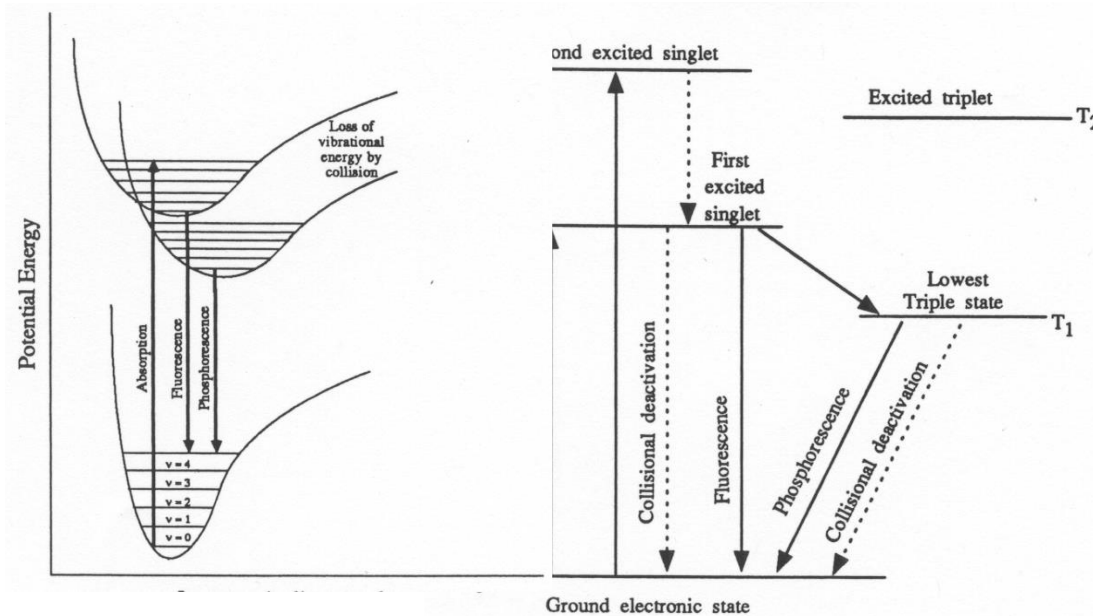
- Rayleigh and Raman scattering occurs essentially instantaneously.
Not allowing other energy conversion phenomena to occur.

Fluorescence and phosphorescence

- Photoluminescence with time delay

Fluorescence

- Emission when the excited from singlet state to ground,
- lifetime is about $10^{-10} \sim 10^{-5}$ s.

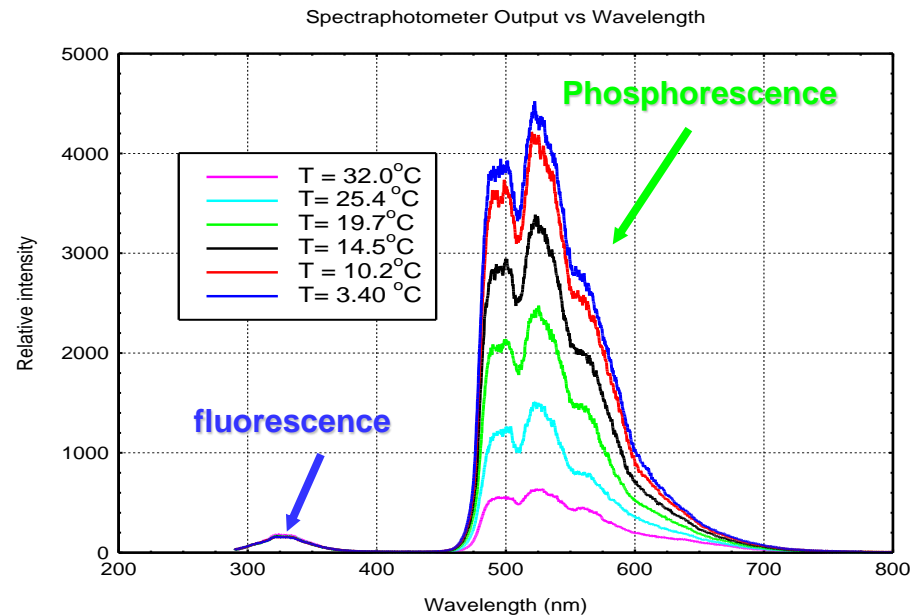
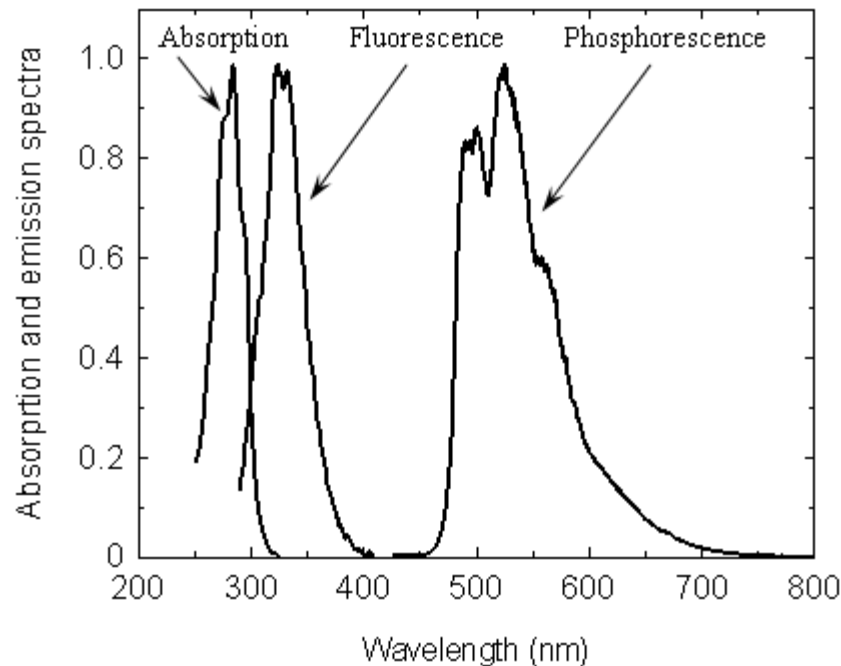


Rhodamine B

□ Fluorescence and phosphorescence

Phosphorescence

- Emission when the excited atom or molecule from triplet state to ground,
- lifetime is about $10^{-4} \sim 10^{-5}$ s.



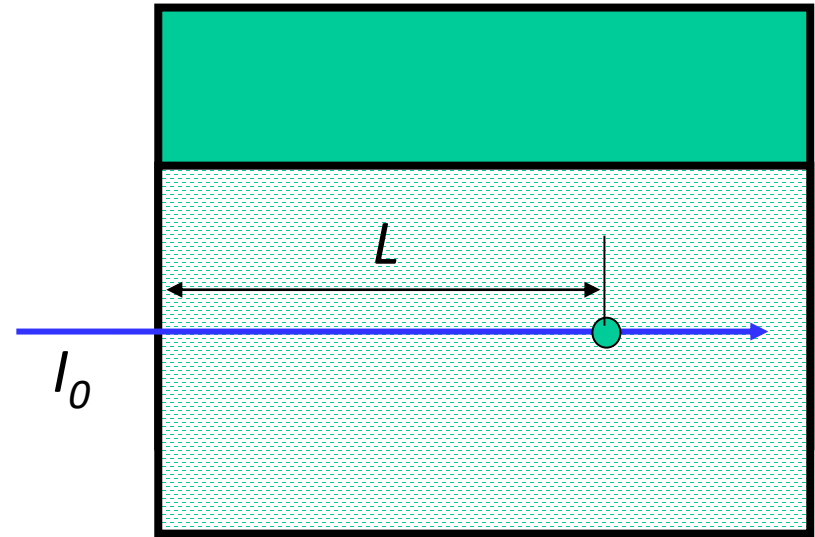
MTV chemical: 1-BrNp•Mβ-CD•ROH complex

□ Absorption

Light is transmitted through a material, it will be absorbed by the molecules of the material

Beer's law: $I = I_0 \exp(-\alpha L)$

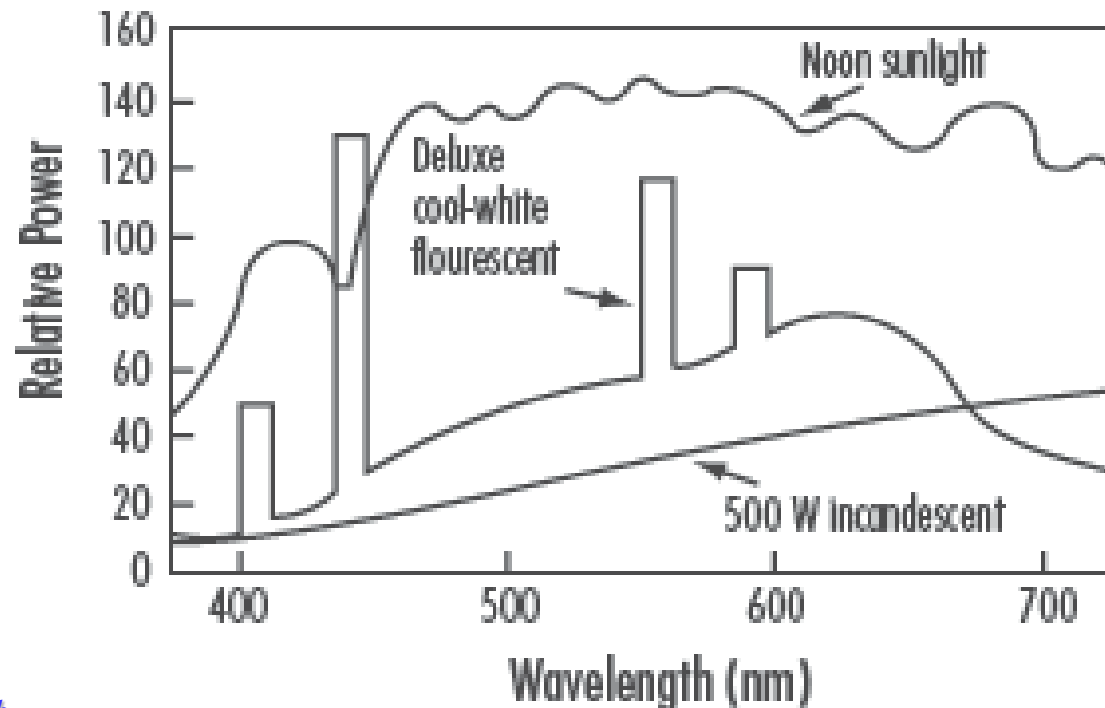
- α is the absorption or attenuation coefficient*
- $L_c = 1/\alpha$ is called penetration depth.*
- When $L = L_c$, $I/I_0 = 1/e = 37\%$, i.e., 63% energy was absorbed*
- Metals have very small $L_c = 1/\alpha$.*
 - Copper, $L_c = 0.6 \text{ nm}$ for 100 nm UV light*
 - Copper, $L_c = 6.0 \text{ nm}$ for 1000 nm infrared light.*
 - 2nm copper plate as a low pass filter.*



ILLUMINATION

Light sources:

- **Thermal source:**
 - Lamps: Continuous wave (CW)
 - Flash lamps (Pulsed)
 - Arc lamps
- **Laser sources**
 - Continuous wave (CW)
 - Pulsed laser
 - Single wavelength

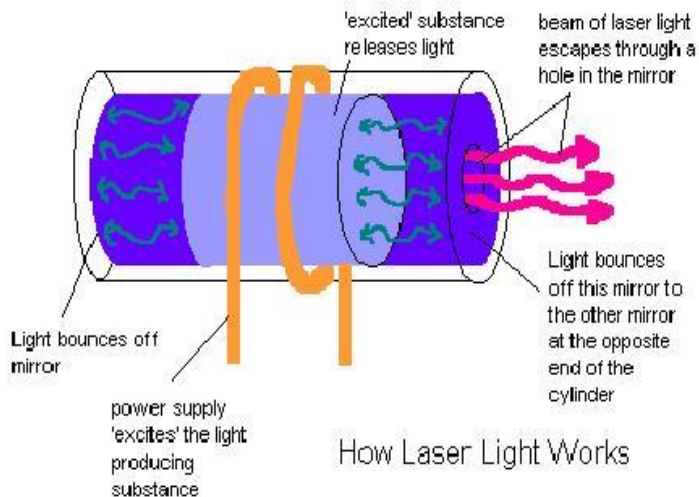
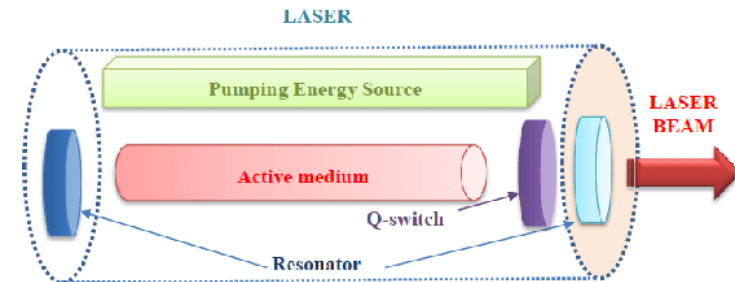


Laser

- *Laser: Light Amplification by Stimulated Emission of Radiation (LASER)*

- *How a laser works:*

- *Radiation energy is produced by an activated medium(can be gas, crystal or semiconductor or liquid solution).*
- *The medium consists of particles (atom, ions or molecules).*
- *When a photo, having energy $h\nu$, approaching the particles, the photo may be absorbed cause an electron or atoms to be raised temporarily to high-energy level.*
- *When the excited electron or molecule to return ground level, spontaneous emission or stimulated emission would take place.*



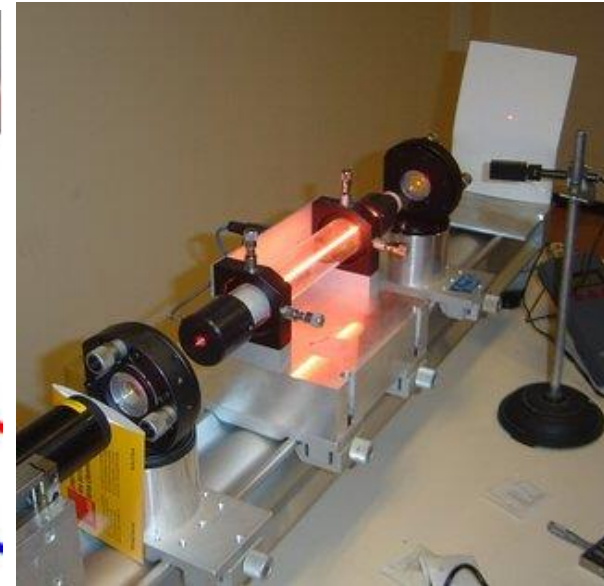
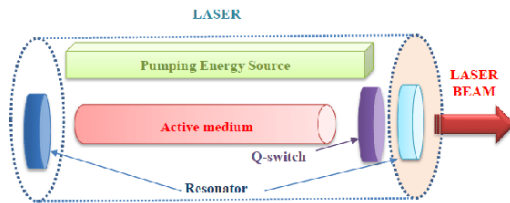
How Laser Light Works



Commonly used Lasers

–Helium-neon (He-Ne) laser

- Active medium is helium neon atoms
- Continuous wave laser
- Power 0.3 ~15 mW
- $\lambda = 633\text{nm}$ (red)



–Argon-ion (Ar-ion) laser

- Active medium is argon atoms maintained at the ion state.
- Continuous wave laser
- Power level: 100 mW ~10 W
- Have seven wavelengths
- $\lambda = 488\text{nm}$ (blue)
- $\lambda = 514.5\text{nm}$ (green)
- LDV application
- LIF in liquid flows



□ Commonly used Lasers

–Nd-YAG laser

- Solid-state laser
- Active medium: neodymium (Nd^{+3}) as active medium incorporated as an impurity into a crystal of Yttrium-Aluminium-Garnet (YAG) as a host
- Flash lamp is used as external source
- pulsed laser: 10 -400mJ/pulse or more
- Pulse duration: 100ps ~ 10ns
- Wavelength of tube $\lambda = 1064\text{nm}$ (infrared)
- SHG: $\lambda = 532\text{nm}$ (green), THG: $\lambda = 355\text{nm}$ (UV), FHG: $\lambda = 266\text{nm}$ (deepUV)
- PIV, MTV, PLIF
- Repetition rate can be as high as 30 Hz.

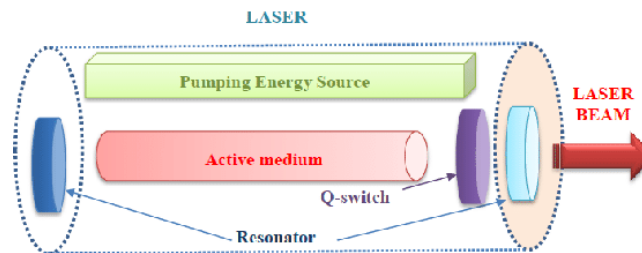


–Dye laser

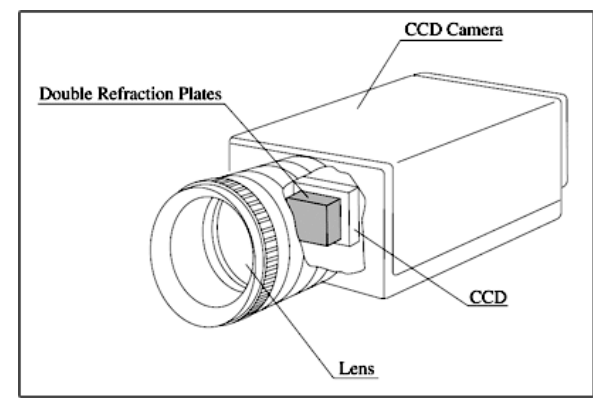
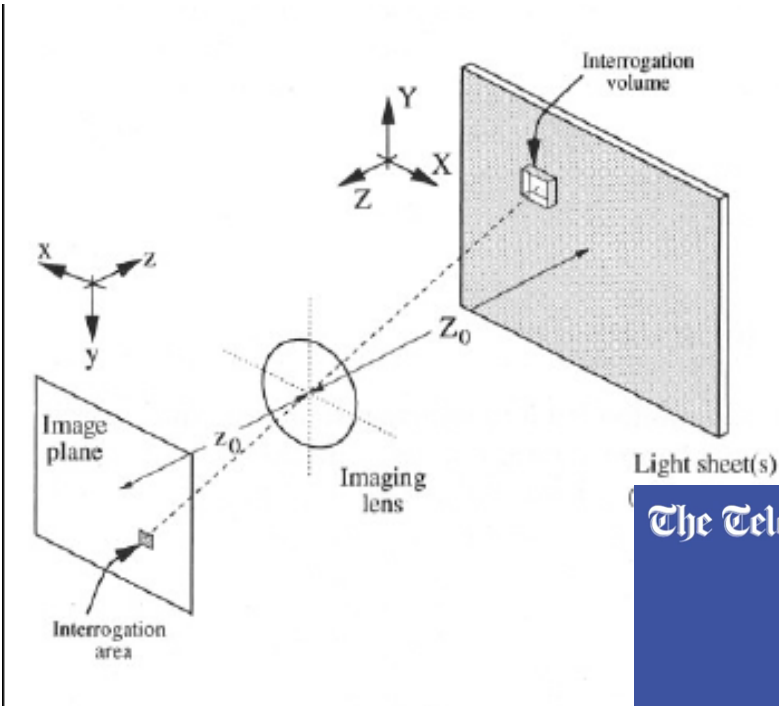
- Active medium: complex multi-atomic organic molecules
- $\lambda = 200\text{nm} \sim 1500\text{nm}$

–Excimer laser

- Gas laser KrF and XeCl
- High-energy
- UV wavelength
- Pulsed laser
- high repetition frequency

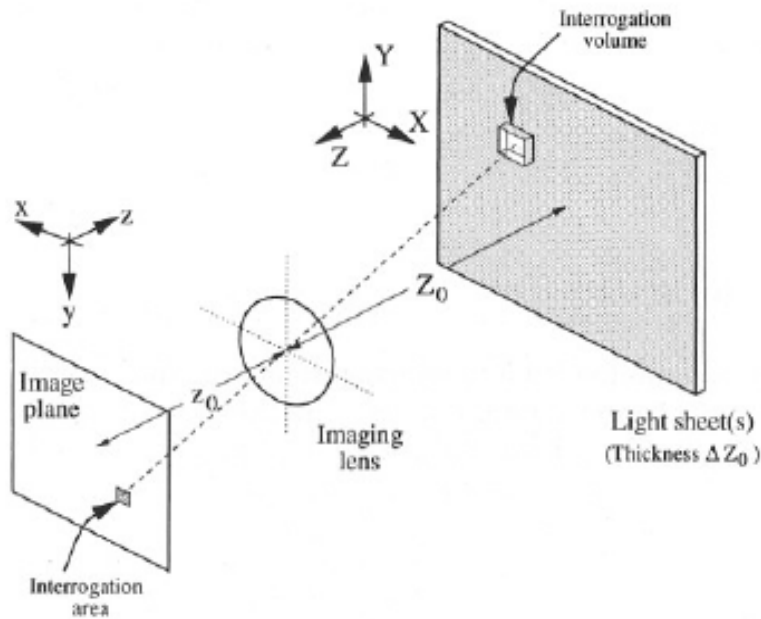


❑ Light sensing and recording



The Telegraph

□ Lenses



- Focal length: f
- $f/\#$, “F-number”: defined as the ratio of focal distance of the lens and its clear aperture diameter.
- Depth of focus $H = 2 \cdot f/\# \cdot c \cdot Z/f$



$f2$ F2.8

wide aperture

more light

small number



$f4$ $f5.6$ $f8$ $f11$



$f16$ $f22$

small aperture

less light

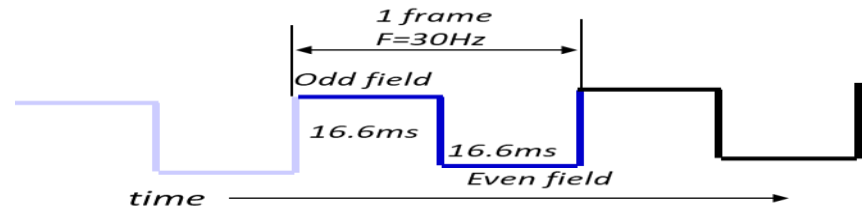
larger number



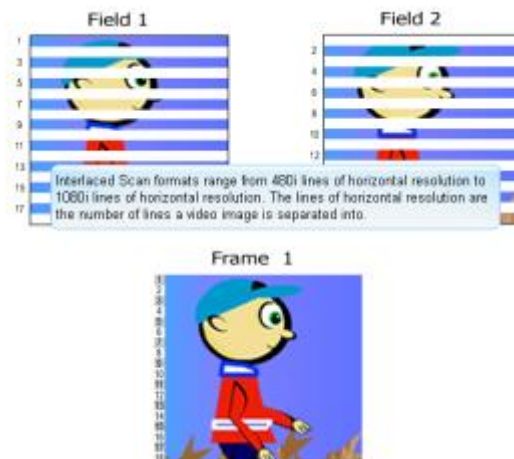
❑ Interlaced Cameras

- The fastest response time of human being for images is about **~ 15Hz**
- Video format:
 - PAL (Phase Alternating Line) format with frame rate of $f=25\text{Hz}$ (sometimes in 50Hz). Used by U.K., Germany, Spain, Portugal, Italy, China, India, most of Africa, and the Middle East
 - National Television Standards Committee (NTSC) with frame rate of $f=30\text{Hz}$. Used by U.S., Canada, Mexico, some parts of Central and South America, Japan, Taiwan, and Korea.

480 pixels by 640 pixels



Interlaced camera



❑ Interlaced camera vs. Progressive Scan Camera

Progressive scan details:



Interlaced scan details:



2CIF details:



Note: In these examples, the cameras have been using the same lens. The car has been driving at 20 km/h (15 mph) using cruise control.

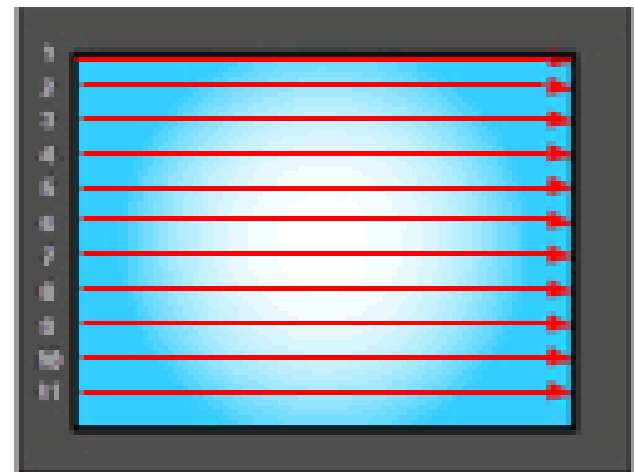


Interlaced



Progressive Scan
(Non-interlaced)

- All image systems produce a clear image of the background
- Jagged edges from motion with interlaced scan
- Motion blur caused by the lack of resolution in the 2CIF sample
- Only progressive scan makes it possible to identify the driver



One complete frame
using progressive scanning

❏ Electronic shutter modes

- **Rolling shutter:** The sensor is exposed line by line. Each of the pixels integrate light for the specified exposure time; however, not all pixels are exposing at the same time. The start time for each pixel's exposure is a function of sensor position. This mode is typical of large format sensors, such as digital SLR cameras.
- **Global shutter:** Each pixel integrates light for the specified exposure time simultaneously. This method is preferred for capturing highly dynamic events. This mode is typical of high-speed CMOS cameras which can operate at frame rates beyond 1 million frames per second.



Rolling shutter
Point Grey Cameras



Global shutter
Point Grey Cameras

❑ Mystery of flying rods



- *WT testing @ 13:30;*
- *High-speed Imaging @29:30; 33:00; 38:20*

